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Answer: B

QUESTION: 600

Which of the following industries did not experience major growth from the late 1800s to the early 1900s?

A. Beef

B. Steel

C. Oil

D. Cloth

E. Wood

Answer: A

QUESTION: 601

Which statement does not accurately portray changes to the American labor force wrought by industrialization in the late 19th and early 20th centuries?

A. By 1880, there were almost five million individuals working in industrial jobs.

B. More than two million women were employed in factories or offices by 1880.

C. By the year 1890, 25% of children aged 10 to 15 years were in the workforce.

D. During these years it was common for workers to sustain injuries and illness.

E. Workers' compensation, disability, and retirement did not exist for workers.

Answer: C

QUESTION: 602

Which of the following is true regarding business consolidation around the turn of the 19th and20th centuries in American industries?

- A. John D Rockefeller used vertical integration with his Standard Oil Company.
- B. Gustavus Swift was the first to use horizontal integration with meat packing.
- C. Social Darwinists were opposed to consolidation as well as to deregulation.
- D. Economists against consolidation favored more control of natural processes.
- E. The Sherman Anti-Trust Act proved quite effective in preventing monopoly.

Answer: D

QUESTION: 603

Which of the following statistics is not accurate with respect to the urbanization of America between 1870 and 1920?

A. The number of Americans who lived in cities grew more than fivefold during this time.

B. By 1920, almost half of the American population lived in cities with over 2500 people.

C. Cities having populations greater than 100 000 went up by more than 3.5 times.

D. Cities having populations greater than 500 000 increased to six-fold.

Answer: B

QUESTION: 604

Around the turn of the 19th and 20th century, when city slums resulted from overcrowding due to immigration, in which of the following areas was the least improvement made?

- A. Housing
- B. Sewage
- C. Firefighting
- D. Street lights
- E. Water supply

Answer: A

QUESTION: 605

Which of the following was not a development of city culture in America around the turn of the 19th and20th century?

- A. Baseball had become America's favorite professional sport by the 1880s.
- B. Vaudeville shows rode trains from cities to small towns across America.
- C. Early directors liked W Griffith developed the motion picture industry.
- D. Pulitzer and Hearst gave mass popularity to newspapers and magazines.

E. All of these were American urban cultural developments during this time.

Answer: E

Section 48: Sec Forty Eight (606 to 614) Details:High Pressure Chemistry Practice

QUESTION: 606

What is the pressure of a mixture of CO_2 , SO_2 and H_20 gases, if each gas has a partial pressure of 250 torr?

A. 250 torr B. 500 torr C. 750 torr D. 1,000 torr

Answer: C

QUESTION: 607

A gas sample has a volume of 25.0 milliliters at a pressure of 1.0 atmosphere. If the volume increases to 50.0 milliliters and the temperature remains constant, the new pressure will be

A. 1.0 atm B. 2.0 atm C. .250 atm D. .500 atm

Answer: D

QUESTION: 608

An increase of the temperature of a system at equilibrium favors the

- A. Endothermic reaction and decreases its rate
- B. Endothermic reaction and increases its rate
- C. Exothermic reaction and decreases its rate
- D. Exothermic reaction and increases its rate

Answer: B

QUESTION: 609

A sample of oxygen gas in a closed system has a volume of 200 milliliters at 600 k. If the pressure is held constant and the temperature is lowered to 300 K, the volume of the gas will be

- A. 100 ml
- B. 200 ml
- C. 300 ml
- D. 400 ml

Answer: A

QUESTION: 610

An assumption of the kinetic energy of gases is that the particles of a gas have

- A. Little attraction for each other and a significant volume
- B. Little attraction for each other and an insignificant volume
- C. Strong attraction for each other and a significant volume
- D. Strong attraction for each other and an insignificant volume

Answer: B

QUESTION: 611

What is the volume occupied by 2.0 moles of Ar(g) at STP?

A. 22.4 L B. 44.8 L C. 89.6 L D. 179 L

Answer: B

QUESTION: 612

Which gas is least likely to obey the ideal gas laws at very high temperatures and very low temperatures?

A. Kr

B. Ne C. He D. Xe

Answer: D

QUESTION: 613

A gas at STP has a volume of 1.0 liters. If the pressure is doubled and the temperature remains constant, the new volume of the gas will be

A. .25 L B. 2.0 L C. .50 L D. 4.0 L

Answer: C

QUESTION: 614

At 1 atmosphere of pressure, 25.0 grams of a compound at its boiling point is converted to a gas by the addition of 8,180 calories. What is the heat of vaporization for this compound, in calories per gram?

A. 25.0 cal/g B. 327 cal/g C. 540 cal/g D. 8,140 cal/g

Answer: B



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